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HYDROCARBON COMPOUNDS SECTION
22.1 HYDROCARBONS (pages 693-701)
This section describes the bonding in
hydrocarbons and distinguishes straight-
chain from branched- chain alkanes. It
also provides rules for naming branch-
chained alkanes.

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Hydrocarbons are organic molecules

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that only contain two elements: carbon and hydrogen. Common hydrocarbons include methane, propane, and butane. Hydrocarbons are the primary constituents of fossil...

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minutes, 16 seconds 1,088 views Part 1
of , hydrocarbons , includes the

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Section 22.1 • Alkyl Halides and Aryl
Halides 787 Table 22.1 Organic
Compounds and Their Functional Groups
Compound Type General Formula
Functional Group Halocarbon R—X (=
F, Cl, Br, I) Halogen Alcohol R—OH
Hydroxyl Ether R—OH ' Ether Amine
R—N H 2 Amino Aldehyde — O *— C — H
Carbonyl Ketone — O R—C Carbonyl
Carboxylic ...

Chapter 22: Substituted Hydrocarbons and Their Reactions

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Answer as skillfully as review them wherever you are now.

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CHAPTER 22, Hydrocarbon Compounds (continued) 8. Circle the letter of each compound that is an aliphatic compound. ()1-butene)acetylene ()2-methylpropane 9. What are the major attractive forces between alkane, alkene, or alkyne molecules? 10. Complete the table below with the names of the indicated alkanes, alkenes, and alkynes.

This section describes the bonding in hydrocarbons and ...

Section 21.1 • Introduction to Hydrocarbons 745 Figure 21.3 Methane—a hydrocar-bon found in natural gas—is the simplest hydrocarbon. Identify In addition to hydrogen, what other elements readily bond with carbon? Organic chemistry Today, the term organic compound is

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applied to all carbon-containing compounds with the primary exceptions of

Chapter 21: Hydrocarbons

Chapter 22: Substituted Hydrocarbons and Their Reactions. A functional group is a (n) The addition of a functional group to a.... Double and triple bonds between two carb.... A halocarbon is a (n) atom or group of atoms that always reacts in a certain way in.... a substance with physical and chemical properties that differ....

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Hydrocarbon Compounds 719 CHAPTER 22 Assessment 37. pentane:

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CH₃CH₂CH₂CH₂CH₃ hexane:
CH₃CH₂CH₂CH₂CH₂CH₃ 38. a. propane
b. octane c. pentane 39. 40. a.
2-methylbutane b. 2,3-dimethylbutane c.
3-methylhexane 41. The carbon-carbon
bonds are non-polar and the carbon-
hydrogen bonds are very weakly polar..
42. a. propene b. trans-2-pentene c.
4-methyl-1-pentene

CHAPTER 22 Study Guide - Quia

Title: 05_Chem_GRSW_Ch22.SE/TE

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05 Chem GRSW Ch22.SE/TE

Chapter 22: Organic Compounds. 22.1
Hydrocarbons; 22.2 Unsaturated
Hydrocarbons; 22.3 Functional Groups;
22.4 Alcohols and Ethers; 22.5 Amines
and Alkaloids; 22.6 Carbonyl
Compounds; 22.7 Polymers; Chapter 23:
The Nutrients of Life. 23.1 Biomolecules
Are Molecules Produced and Used by
Organisms; 23.2 Carbohydrates Give
Structure and Energy

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22.2 Unsaturated Hydrocarbons | Conceptual Academy

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Chapter 22 - Hydrocarbon Compounds - 22.3 Isomers - 22.3 ...

Chapter 22 Hydrocarbon Compounds
Name 12, 13. ST 14. 150 Straight-chain
alkanes contain 10 carbon atoms. A
substituent can take the place of a
hydrogen atom on a parent hydrocarbon
molecule.

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Chapter 22 - Hydrocarbon Compounds - 22.2 Unsaturated ...

Read Ch. 22 section 1 (pp. 693-701) and complete the activities below: On page 696 construct Table 22.2 for hexane: FORMULA DESCRIPTION On page 697 read conceptual problem 22.1 and answer the practice problems. 1.) Chapter 22, section 1

Chapter 22, section 1 - SharpSchool

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